

A NOVEL ROUTE TO SUBSTITUTED PHOSPHONATES VIA CONJUGATE ADDITION OF ORGANOMETALLICS TO 1-(FUNCTIONALLY)SUBSTITUTED ALKENE PHOSPHONATES

F. BARBOT, E. PARAISSO and Ph. MIGINIAC

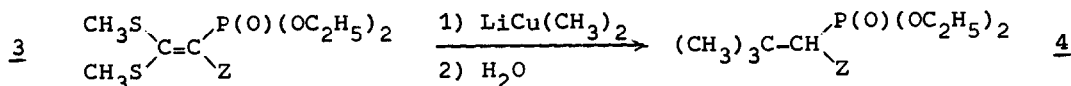
Laboratoire de Chimie des Organométalliques. Groupe de Recherches de Chimie Organique, Université de Poitiers, 86022 POITIERS, France.

Summary : Organometallics readily add to the double bond of 1-(functionally) substituted alkene phosphonates to give highly substituted phosphonates.

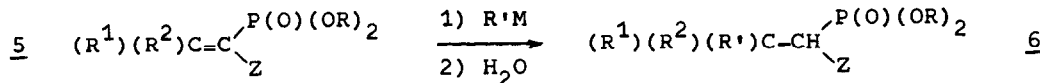
Whereas the conjugate addition of organometallics to alkyldiene malonic esters 1 is a well known reaction (1-3), it appears from the literature that such a conjugate addition has not been described with 1-(functionally) substituted alkene phosphonates 2 (Z= CN, COOR', CO-CH₃).



During the course of a study on the reactivity of activated double bonds towards organometallics, we therefore decided to consider the case of type 2 ethylenic compounds. A recent publication (4) dealing with the formation of 4 from 3 and lithium dimethylcuprate, prompts us to report the first results we have obtained so far.

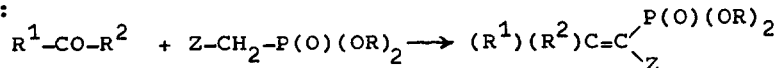


The reaction of saturated and allylic lithium organocuprates, methylmagnesium iodide and allylzinc bromide with the 1-(functionally) substituted alkene phosphonates 5 gives with a good yield the phosphonates 6 arising from a conjugate addition (Table)(5).

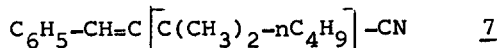


This reaction is an easy way to substituted phosphonates which cannot be prepared easily by using the Michaelis-Arbuzov rearrangement (6).

The starting alkene phosphonates were prepared by using the following reaction (7-10) :



Two of the phosphonates 6 we have obtained have been used in a Wittig-Horner reaction. Thus, the phosphonate 6b, reacted with sodium hydride in tetrahydrofuran and then with benzaldehyde, gave the 2-substituted cinnamonnitrile 7 (Yield : 50%).



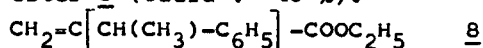
Table

$$\underline{5} \quad (R^1)(R^2)C=C \begin{array}{l} \nearrow P(O)(OR)_2 \\ \searrow Z \end{array} \xrightarrow[2) H_2O]{1) R^*M} (R^1)(R^2)(R^*)C-CH \begin{array}{l} \nearrow P(O)(OR)_2 \\ \searrow Z \end{array} \quad \underline{6}$$

Entry	R*M	R ¹	R ²	R	Z	temp.°C	Yield %
a	LiCu(CH ₃) ₂ [*]	C ₆ H ₅	H	CH ₃	CN	-30 to 20	90
b	LiCu(nC ₄ H ₉) ₂ ^{**}	CH ₃	CH ₃	C ₂ H ₅	CN	-30 to 20	80
c	LiCu(nC ₄ H ₉) ₂ ^{**}	C ₂ H ₅	H	C ₂ H ₅	CN	-30 to 20	60
d	LiCu(CH ₃) ₂ [*]	C ₆ H ₅	H	C ₂ H ₅	COOC ₂ H ₅	-30 to 20	90
e	CH ₃ MgI [*]	C ₆ H ₅	H	C ₂ H ₅	COOC ₂ H ₅	20	89
f	BrZnCH ₂ -CH=CH ₂ ^{***}	C ₆ H ₅	H	C ₂ H ₅	COOC ₂ H ₅	20	73
g	LiCu(CH ₃) ₂ [*]	C ₆ H ₅	H	C ₂ H ₅	CO-CH ₃	-30 to 20	82
h	LiCu(CH ₂ -CH=CH ₂) ₂ ^{****}	C ₆ H ₅	H	C ₂ H ₅	CO-CH ₃	-30 to 20	75

Solvent: * ether; ** ether/hexane; *** tetrahydrofuran; **** ether/tetrahydrofuran.

The reaction of the phosphonate 6de with an excess of formaldehyde (30 % aqueous solution) in the presence of a saturated solution of potassium carbonate, according to (11) but using more drastic conditions (72h/90°C), gave the 2-substituted acrylic ester 8 (Yield : 45 %).



Generalisation of this work is now under current investigation .

References

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- (5) All compounds obtained in this work showed satisfactory analytical data; yields, based on 5 (0.03 mole) are not optimized and refer to distilled products. ¹H NMR (CCl₄): 6a (250MHz): 1.5-1.55 (2d, 3H, CH₃-C, J7Hz); 3.35-3.55 (m, 1H, C₆H₅-CH); 3.6-3.9 (2dd, 1H, CH-CN, J(H-H)5Hz, J(H-P)22Hz); 3.5-3.85 (4d, 6H, OCH₃, J(H-P)10.7Hz). 6b (60MHz): 0.9 (t, 3H, CH₃-C-C, J6Hz); 1.2 (s, 6H, C(CH₃)₂); 1.1-1.7 (m, 12H, (CH₂)₃/O-C-CH₃); 2.95 (d, 1H, CH-CN, J24Hz); 3.9-4.5 (m, 4H, OCH₂). 6c (60MHz): 0.8-1.1 (m, 6H, CH₃-C-C); 1.1-1.8 (m, 15H, CH₂-CH-(CH₂)₃/O-C-CH₃); 2.05 (dd, 1H, CH-CN, J(H-H)4Hz, J(H-P)25Hz); 3.9-4.5 (m, 4H, OCH₂). 6d/e (500MHz): 0.9-1.4 (6t, 6H, CH₃-CH₂, J7Hz); 1.47/1.33 (2d, 3H, CH₃-CH, J7Hz); 3.20-3.26 (2dd, 1H (CH-P(O), J(H-P)20Hz); 3.4-3.5 (m, 1H, CH-CH₃); 3.8-4.3 (2m, 6H, OCH₂). 6f (250MHz): 0.88/1.28 (2t, CO-O-C-CH₃, J6.5Hz); 1.02/1.07/1.31/1.32 (4dt, P-O-C-CH₃, J(H-H)6.5Hz, J(HC-C-O-P)0.5Hz); 4.76-5.7 (m, 3H, CH=CH₂); 7.15-7.40 (m, 5H, C₆H₅). 6g (60MHz): 0.9-1.6 (m, 9H, CH₃-CH/CH₂-CH₂); 1.85/2.25 (2s, 3H, CH₃-CO); 3.3-4.4 (m, 6H, OCH₂/CH-CN); 7-7.4 (m, 5H, C₆H₅). 6h (60MHz): 0.9-1.5 (m, 6H, O-C-CH₃); 1.85/2.30 (2s, 3H, CH₃-CO); 2.2-2.6 (m, 2H, CH₂-C=); 3.3-4.5 (m, 6H, OCH₂/CH-CH); 4.6-5.8 (m, 3H, CH=CH₂); 7.1-7.5 (m, 5H, C₆H₅).
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